

Enantioselective Synthesis of Polyfunctional Small Building Blocks with a Quaternary Stereogenic Center

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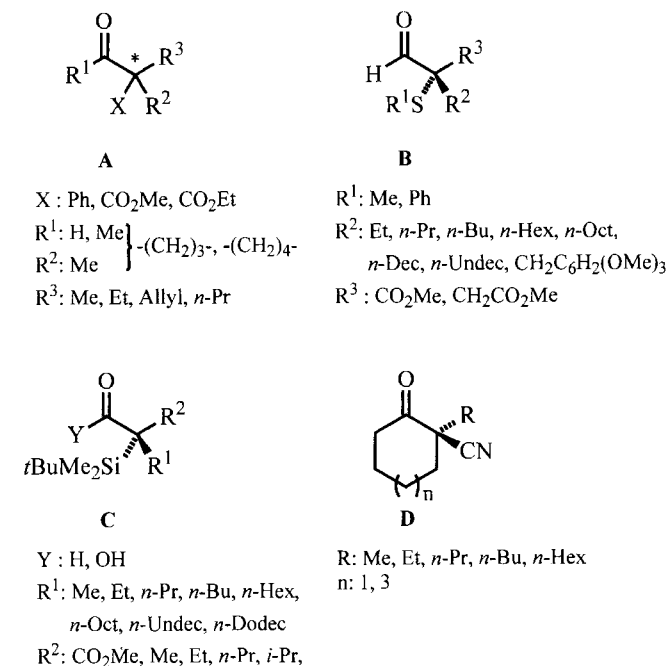
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Application of the SAMP-/RAMP-hydrazone method offers an efficient and flexible access to compounds with a quaternary stereogenic center. Examples bearing silyl, thio, phenyl, aldo, keto, ester, and alkyl functionalities are described. The 2-phenylaldehydes and -ketones and the β -keto esters **4** were obtained in good overall yields and with variable enantiomeric excesses. The synthesis of the thiolated 2-formyl and 3-formyl carboxylic esters **12** was achieved in high overall

yields and with high enantiomeric excesses. The silylated carboxylic ester and acids **16** were produced in moderate to high overall yields and with moderate to excellent enantiomeric excesses, depending on the electrophile used for the quaternization. The absolute configurations of the compounds **12** and **16** were determined by X-ray structure analyses, and the mechanism of the quaternization is postulated.

The design of asymmetric syntheses for molecules with a quaternary carbon center as chiral centers for the generation of natural products and biologically active compounds is an attractive and challenging field in synthetic organic chemistry. Since Martin's review^[1] of the creation of quaternary carbon centers a number of efforts have been made during the last decade to extend the range of enantioselective methods^[2]. Since highly regio-, diastereo-, and enantioselective electrophilic substitutions at the C_α position of the carbonyl group of ketones and aldehydes by the SAMP-/RAMP-hydrazone method^[3] are well established, it was decided to explore the high potential of this methodology in the asymmetric creation of small polyfunctional building blocks bearing a quaternary carbon center^[3a,4]. Our research group has already reported on a highly enantioselective synthesis of 2-alkyl-2-cyanocycloalkanones **D** with a quaternary stereogenic center^[5] and demonstrated in a short communication^[6] an efficient enantioselective route to quaternary 2-formyl esters and carboxylic acids. The enantioselective synthesis of the small polyfunctional building blocks **A**, **B**, and **C** using the SAMP-/RAMP-hydrazone method is described in this paper in full detail.

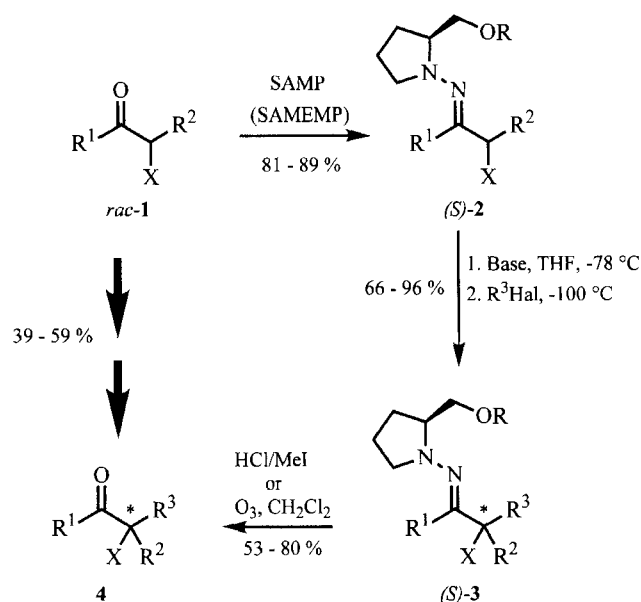
The racemic 2-phenyl aldehydes and ketones and the racemic 2-oxo carboxylic acids were easily transformed into the corresponding SAMP- and the (*S*)-1-amino-2-[(methoxyethoxy)methoxy]methylpyrrolidine (SAMEMP)-hydrazones **2** (for the synthesis of SEMEMP see scheme 2). The β -keto ester SAMP-hydrazones exist as a 1:2 mixture of the hydrazone/ene hydrazone tautomeric form. The 2-phenyl derivatives **2a–c** were added to a solution of LDA in THF at -78°C and after alkylation with alkyl halides at -100°C the 2-phenylhydrazones **3a–d** were prepared in



good yields. Metalation of the β -keto ester SAMP- and the SAMEMP-hydrazones^[3a,4] with *n*BuLi in THF at -100°C followed by trapping of the intermediate aza-allyl anion with alkyl halides furnished the corresponding hydrazones **3e–l** in good yields. After hydrazone cleavage (**3a, b**: acidic hydrolysis via the methiodide in a two-phase system; **3c, d**: acidic hydrolysis in a two-phase system; **3d–l**: ozonolysis) the optically active 2-phenyl aldehydes and ketones and the β -keto esters **4** were obtained in good yields (53–80%) with

enantiomeric excesses ranging between 18 and 93% (Scheme 1).

Scheme 1

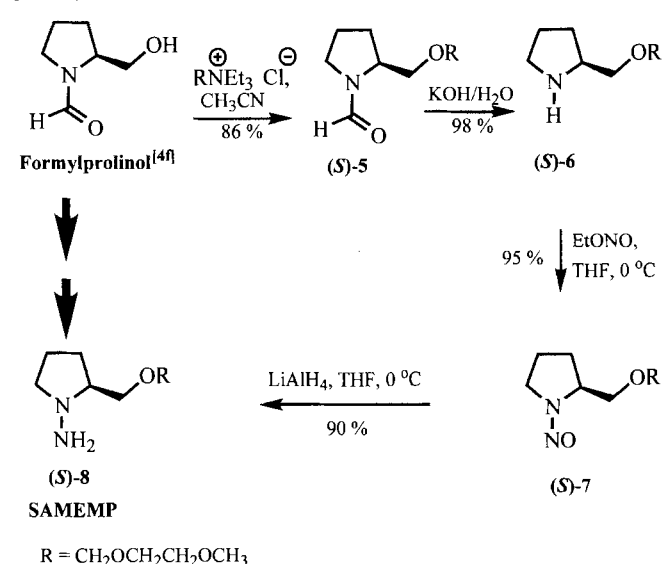


X: Ph, CO₂Me, CO₂Et
 R: CH₃, CH₂OCH₂CH₂OCH₃
 R¹: H, Me } -(CH₂)₃-, -(CH₂)₄-
 R²: Me }
 R³: Me, Et, Allyl, *n*-Pr

The diastereomeric and the enantiomeric excesses of the hydrazones **3** and the quaternary carbonyl compounds **4**

were determined by LIS-NMR techniques using the achiral shift reagent Eu(fod)₃ and the chiral shift reagent Eu(hfc)₃, respectively. The absolute configuration of the hydrazones **3** were not determined.

Scheme 2



SAMEMP **8** was used as a chiral auxiliary in an attempt to achieve a stronger chelation by the increased number of oxygens in the pyrrolidine side chain. A comparison of the results (**3a** versus **3k**, **3b** versus **3l**) revealed no significant effect on the degree of asymmetric induction (Table 1). Although the ee values of the quaternary carbonyl compounds **4** are in general relatively low (18–93%) the excep-

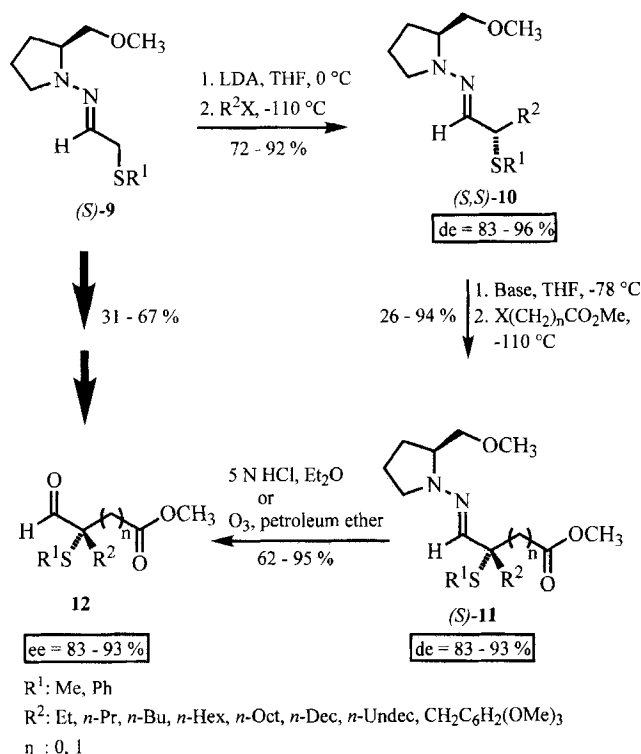
Table 1. SAMP- and SAMEMP-hydrazones **3** and α -quaternary carbonyl compounds **4**

2[a]	R ¹	R ²	X	3[a]	R ³	yield [%]	α_D^{RT} neat or (c, C ₆ H ₆)	de ^[b] [%]	4[a]	yield [%]	$[\alpha]_D^{RT}$ (c, CHCl ₃)	ee ^[c] [%]
(S)-2a	H	Me	Ph	(S)-3a	Allyl	85	-76.6	30	4a	64	-35.2 (1.39)	30
(S)-2b	H	Me	Ph	(S)-3b	Et	83	-78.9	28	4b ^[d]	76	-2.9 (4.19)	25
(S)-2c	-(CH ₂) ₃ -		Ph	(S)-3c	Me	69	+289.5 (1.19)	[e]	4c ^[d]	73	+69.5 (1.24) ^[f]	77
(S)-2d	-(CH ₂) ₄ -		Ph	(S)-3d	Me	66	+402.8 (1.42)	[e]	4d ^[d]	80	+200.4(2.38) ^[g]	93
(S)-2e	-(CH ₂) ₃ -	CO ₂ Me		(S)-3e	Me	85	+240.2	60	4e	78	-7.4 (1.16)	60
(S)-2f	-(CH ₂) ₃ -	CO ₂ Me		(S)-3f	Et	96	+233.5	23	4f	73	+3.7 (1.21)	18
(S)-2g	-(CH ₂) ₄ -	CO ₂ Me		(S)-3g	Et	93	+211.6	42	4g	68	+67.6 (3.04)	43
(S)-2h	Me	Me	CO ₂ Et	(S)-3h	Et	85	+232.7	[e]	4h	62	+1.8 (1.33)	27
(S)-2i	-(CH ₂) ₄ -	CO ₂ Et		(S)-3i	Allyl	73	+163.8	20	—	—	—	—
(S)-2j ^[h]	Me	Me	CO ₂ Et	(S)-3j ^[h]	<i>n</i> -Pr	83 ^[i]	+193.5	[e]	4i ^[h]	76	+2.1 (1.46)	31
(S)-2k ^[h]	H	Me	Ph	(S)-3k ^[h]	Allyl	86	-62.5	[e]	4a ^[h]	76	-27.6 (1.17)	24
(S)-2l ^[h]	H	Me	Ph	(S)-3l ^[h]	Et	90	-66.0	[e]	4b ^[d,h]	53	-3.2 (1.36)	28

^[a] The absolute configuration was not determined. — ^[b] Determined by Eu(fod)₃ shift. — ^[c] Determined by Eu(hfc)₃ shift. — ^[d] The absolute configuration was determined to be (*R*) by a comparison with literature data^[16,17]. — ^[e] Not determined. — ^[f] Measured in CH₃CH₂OH. — ^[g] Measured in C₆H₁₂. — ^[h] SAMEMP [(S)-8] was used instead of SAMP as auxiliary [R = CH₂OCH₂CH₂OCH₃]. — ^[i] 50% of the substance was starting material.

tional case of (*R*)-**4d** (ee = 93%) demonstrated the possibility of high induction by using the SAMP-/RAMP-hydrazone method. This prompted us to optimize the results of this method not only by a variation of the reaction parameters (temperature, solvent, cosolvent, base) but also by changing the neighboring anion-stabilizing groups. Instead of phenyl and ester groups, thio, silyl, and cyano^[5] groups were investigated.

Scheme 3

Table 2. SAMP- and RAMP-hydrazones **11** and thiolated 2-formyl and 3-formyl methyl esters **12**

11	R^1	R^2	n	yield [%]	α_{D}^{25} or (c, C_6H_6)	de ^[a] [%]	12	yield [%]	$[\alpha]_{\text{D}}^{25}$ (c, C_6H_6)	ee ^[b] [%]
(<i>S,R</i>)- 11a	Ph	Et	0	91	-22.9 (1.13)	90	(<i>R</i>)- 12a	90	+155.6 (1.06)	90
(<i>R,S</i>)- 11a ^[c]	Ph	Et	0	69	+23.3 (1.00)	92	(<i>S</i>)- 12a ^[c]	95	-161.7 (1.03)	92
(<i>S,R</i>)- 11b	Ph	<i>n</i> -Pr	0	91	-27.1	92	(<i>R</i>)- 12b	80	+127.8 (0.98)	92
(<i>S,R</i>)- 11c	Ph	<i>n</i> -Bu	0	90	-25.3	92	(<i>R</i>)- 12c	83	+162.7 (1.10)	92 ^[d]
(<i>R,S</i>)- 11c ^[c]	Ph	<i>n</i> -Bu	0	73	+9.9 (0.87)	92	(<i>S</i>)- 12c ^[c]	75	-158.5 (1.05)	92
(<i>S,R</i>)- 11d	Ph	<i>n</i> -Hex	0	85	-11.7 (1.06)	92	(<i>R</i>)- 12d	95	+136.1 (0.98)	92
(<i>S,R</i>)- 11e	Ph	<i>n</i> -Oct	0	77	-19.2	92	(<i>R</i>)- 12e	95	+112.7 (1.15)	92
(<i>S,R</i>)- 11f	Ph	<i>n</i> -Dec	0	94	-14.4	91	(<i>R</i>)- 12f	83	+100.8 (1.07)	91
(<i>S,R</i>)- 11g	Ph	<i>n</i> -Undec	0	91	-14.5	93	(<i>R</i>)- 12g	82	+109.4 (1.20)	93
(<i>S,R</i>)- 11h	Ph	[e]	0	60	-26.3 (1.14)	93	(<i>R</i>)- 12h	62	+141.9 (0.94)	93
(<i>S,R</i>)- 11i	Me	Et	0	82	-85.0	88	(<i>R</i>)- 12i	79	+109.9 (1.38)	88
(<i>S,S</i>)- 11j	Me	Et	1	26	-79.3 (1.13)	91	(<i>S</i>)- 12j	80	-20.9 (0.23)	91
(<i>S,S</i>)- 11k	Ph	Et	1	75	-31.1 (0.97)	83	(<i>S</i>)- 12k	63	-62.6 (1.01)	83
(<i>S,S</i>)- 11l	Ph	<i>n</i> -Hex	1	64	-15.9 (1.06)	87	(<i>S</i>)- 12l	25 ^[f]	-57.5 (1.05)	87

[a] Determined by ^{13}C -NMR spectroscopy. – [b] Indirectly determined by ^{13}C -NMR spectroscopy of the hydrazone **11**. – [c] RAMP was used instead of SAMP as auxiliary. – [d] Determined by ^{19}F -NMR spectroscopy of the Mosher ester^[8] of the corresponding alcohol. – [e] R^2 : $\text{CH}_2\text{C}_6\text{H}_2(\text{OMe})_3$. – [f] Byproduct methyl 3-formylnon-3-enoate.

The α -thiolated hydrazones **9**^[7], easily prepared from the corresponding aldehyde and SAMP, were metalated with LDA in THF at 0°C and the lithio derivatives converted diastereoselectively to the hydrazones **10**^[7] (de = 83–96%) in high yields (72–92%) by alkylation with various electrophiles at -110°C . Deprotonation with a base (*n*BuLi, LDA, or MeLi) in THF at -78°C followed by quaternization of the azaallyl anion with methyl chloroformate or methyl bromoacetate at -110°C yielded the α -thiolated hydrazones **11**. Acylation with methyl chloroformate furnished the 2-formyl ester derivatives **11a–i** in good yields (60–94%) and with high diastereomeric excesses (de = 88–93%). The quaternization with methyl bromoacetate afforded the hydrazones **11j–l** with good diastereomeric excesses (de = 83–91%) but only in moderate yields (26–75%) due to the lower reactivity of methyl bromoacetate compared with methyl chloroformate. The hydrazones **11a–i** were hydrolytically cleaved with 5 N HCl in a two-phase system with diethyl ether to the enantiomerically enriched 2- and 3-formyl esters **12a–i** (ee = 88–93%) in good yields (62–95%). Since the hydrolytic cleavage of **11l** led to the formation of methyl 3-formylnon-3-enoate as a byproduct, resulting in a low yield (25%) of the desired product **12k**, the hydrazone **11k** was oxidatively cleaved with ozone. The antipodes of the *R* enantiomers of the 2-formyl esters were synthesized by using RAMP instead of SAMP as a chiral auxiliary [(*R*)-**12a** versus (*S*)-**12a** and (*R*)-**12c** versus (*S*)-**12c**].

The diastereomeric excesses of hydrazones **10** and **11** were determined by ^{13}C -NMR spectroscopy. The enantiomeric excesses of the 2- and 3-formyl esters **12** were determined indirectly by ^{13}C -NMR spectroscopy of the hydrazones **11**. The hydrolytic cleavage of the quaternary hydrazones took place without racemization, as proven by the determination of the enantiomeric excess of (*R*)-**12c** by ^{19}F -NMR spectroscopy of the 3,3,3-trifluoro-2-methoxy-2-phenylpropionic acid ester (Mosher's ester^[8]) of the corresponding alcohol, which was obtained by reduction of (*R*)-**12c** with NaBH_4 in ethanol.

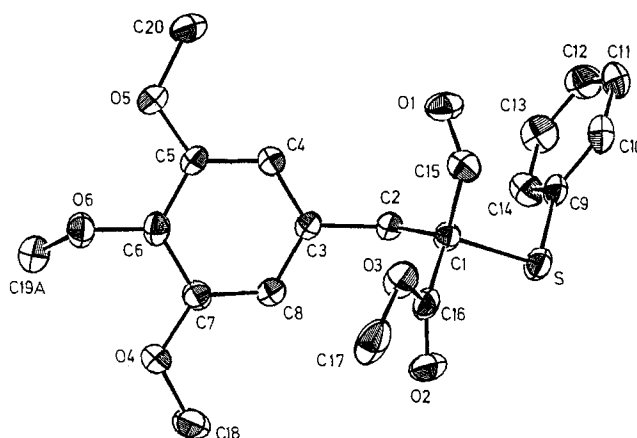


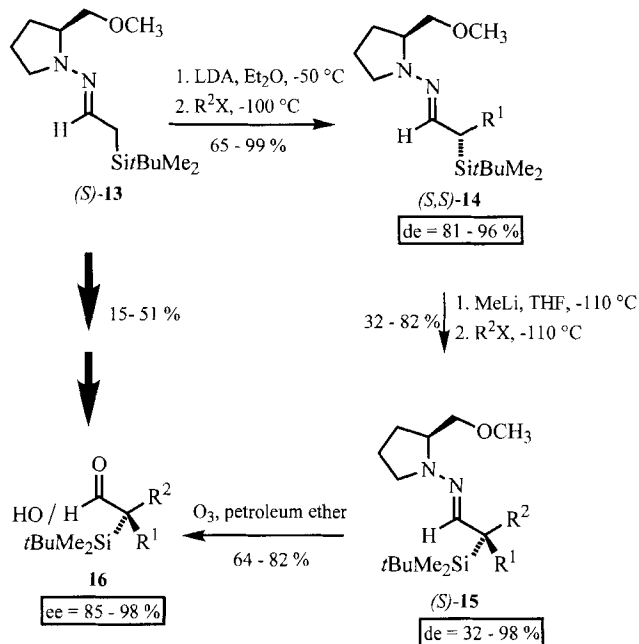
Figure 1. Molecular structure of (*R*)-**12g** in the solid state^[9]. Only one component of the disordered carbon atom C19 is shown (ORTEP^[10] plot)

Since all 2- and 3-formyl esters **12** were previously unknown in optically pure form, the absolute configuration

needed to be determined. Suitable crystals of (*R*)-**12g** for the X-ray structure analysis were obtained from diethyl ether/petroleum ether (1:2) at room temperature (Figure 1). The *R* configuration found at the quaternary stereogenic center confirmed the postulated mechanism for electrophilic substitutions via SAMP-/RAMP-hydrazones^[3a,11].

The α -silylated hydrazone **13**^[6] was prepared by silylation of the deprotonated acetaldehyde SAMP-hydrazone with *tert*-butyldimethylsilyl chloride. Metalation of this hydrazone with LDA in diethyl ether at -50°C and subsequent reaction with different alkyl iodides at -100°C afforded the α -alkylated α -silylated hydrazones **14** in good yields (65–96%) and with high enantiomeric excesses (81–96%). After deprotonation of **14** with MeLi and quaternization with various electrophiles, the hydrazones **15** were obtained with moderate to excellent diastereomeric excesses (*de* = 32–98%) and in good yields (32–82%). After oxidative cleavage with ozone, the 2-formyl esters **16a–f** were obtained in good yields (66–82%) and with high enantiomeric excesses (*ee* = 90–98%). Ozonolysis of hydrazones **15g, j, l, and m** (R^2 : alkyl) gave the corresponding carboxylic acids instead of the aldehydes. The α -quaternary carboxylic acids **16g–j** were obtained in good yields (64–74%) and with high enantiomeric excesses (85–93%).

Scheme 4



R^1 : Me, Et, *n*-Pr, *n*-Bu, *n*-Hex, *n*-Oct, *n*-Undec, *n*-Dodec

R^2 : CO₂Me, Me, Et, *n*-Pr, *i*-Pr

The highest diastereomeric excesses for the quaternary hydrazones **15** were obtained by treatment with methyl chloroformate as the electrophile (*de* = 90–98%), but only moderate yields were achieved (32–62%) because of a competing *N*-acylation of the metalated species, providing the corresponding methyl *N*-[2-(*tert*-butyldimethylsilyl)-1-alkenyl]-*N*-SMP-carbamate. Concerning quaternization with alkyl iodides, the best results were obtained with ethyl iodide (Table 3). Using starting materials with a short side

Table 3. SAMP- and RAMP-hydrazones **15** and silylated quaternary carbonyl compounds **16**

15	R ¹	R ²	yield [%]	α_D^{RT} neat or (c. C ₆ H ₆)	<i>de</i> ^[a] [%]	16	yield [%]	$[\alpha]_D^{RT}$ (c. C ₆ H ₆)	<i>ee</i> ^[b] [%]
(<i>S,R</i>)- 15a	Et	CO ₂ Me	56	+7.1 (1.48)	90	(<i>R</i>)- 16a	77	-183.1 (1.00)	90
(<i>S,R</i>)- 15b	<i>n</i> -Pr	CO ₂ Me	47	-12.0	96	(<i>R</i>)- 16b	66	-167.9 (0.97)	96
(<i>S,R</i>)- 15c	<i>n</i> -Bu	CO ₂ Me	62	+7.1 (0.98)	97	(<i>R</i>)- 16c	82	-124.9 (1.18)	97
(<i>S,R</i>)- 15d	<i>n</i> -Hex	CO ₂ Me	52	0.0 (0.95)	98	(<i>R</i>)- 16d	75	-126.1 (0.93)	98
(<i>S,R</i>)- 15e	<i>n</i> -Oct	CO ₂ Me	43	+4.3 (1.20)	96	(<i>R</i>)- 16e	75	-131.1 (0.97)	96
(<i>R,S</i>)- 15e ^[c]	<i>n</i> -Oct	CO ₂ Me	32	-5.6 (1.18)	98	(<i>S</i>)- 16e ^[c]	73	+107.6 (1.22)	98
(<i>S,R</i>)- 15f	<i>n</i> -Undec	CO ₂ Me	62	+4.1 (1.02)	96	(<i>R</i>)- 16f	74	-112.1 (1.00)	96
(<i>S,S</i>)- 15g	Me	Et	82	-120.5 (1.00)	93	(<i>S</i>)- 16g	70 ^[d]	+2.3 (0.97)	93
(<i>S,R</i>)- 15g	Et	Me	66	-68.7 (1.07)	32	—	—	—	—
(<i>S,R</i>)- 15h	<i>n</i> -Bu	Me	63	-55.6 (1.18)	48	—	—	—	—
(<i>S,R</i>)- 15i	<i>n</i> -Pr	Et	87	-75.0 (1.09)	86	—	—	—	—
(<i>S,S</i>)- 15i	Et	<i>n</i> -Pr	66	-96.6 (1.04)	79	—	—	—	—
(<i>S,R</i>)- 15j	<i>n</i> -Bu	Et	79	-78.8 (1.03)	85	(<i>R</i>)- 16h	74 ^[d]	+9.0 (0.92)	85
(<i>S,S</i>)- 15j	Et	<i>n</i> -Bu	69	-91.7 (1.03)	75	—	—	—	—
(<i>S,R</i>)- 15k	<i>n</i> -Oct	Et	39	-61.3 (0.97)	84	—	—	—	—
(<i>S,R</i>)- 15l	<i>n</i> -Dodec	Et	41	-57.4 (1.03)	88	(<i>R</i>)- 16i	64 ^[d]	+11.6 (1.09)	88
(<i>S,S</i>)- 15m	<i>n</i> -Dodec	<i>i</i> -Pr	41	-85.7 (1.05)	93	(<i>S</i>)- 16j	66 ^[d]	+8.7 (0.98)	93

^[a] Determined by ¹³C-NMR spectroscopy. — ^[b] Indirectly determined by ¹³C-NMR spectroscopy of the hydrazone **15**. — ^[c] RAMP was used instead of SAMP as auxiliary. — ^[d] The corresponding carboxylic acid was isolated instead of the aldehyde after ozonolysis.

chain such as **14a, c, and d**, we synthesized the hydrazones **15g, i, and j** in good yields (79–82%) and with high diastereomeric excesses (*de* = 85–93%). If starting materials with a long side chain were used for the alkylation with ethyl iodide the resultant yields were lower (39–41%), whereas the diastereomeric excesses were in a similar range (*de* = 84–88%). In comparison with ethyl iodide, the use of the other homologues (propyl and methyl iodide) and the branched homologue (isopropyl iodide) resulted in lower chemical yields (39–41%), but with still high asymmetric inductions (*de* = 84–93%). Quaternization with methyl iodide proceeded only in moderate yields (32–48%) with low diastereomeric excesses (*de* = 63–66%). The diastereomeric excesses of the hydrazones **14** and **15** were determined by ¹³C-NMR spectroscopy. The enantiomeric excesses of the 2-formyl ester and carboxylic acids **16** were determined indirectly by ¹³C-NMR spectroscopy of the hydrazones **15**. Racemization by oxidative cleavage with ozone was not observed^[3].

The absolute configurations of the 2-formyl esters and carboxylic acids **16** were determined by X-ray structure analysis of the carboxylic acid (*S*)-**16g** (Figure 2). The *S* configuration found at the new stereogenic center was in agreement with that predicted by the postulated mechanism for electrophilic substitutions via SAMP-/RAMP-hydrazones^[3a,11].

In conclusion, the application of the SAMP-/RAMP-hydrazone method to asymmetric syntheses of highly enantiomerically enriched polyfunctional chirons bearing a quaternary stereogenic center has been shown. The absolute configuration of **B** and **C** and the mechanism of the electro-

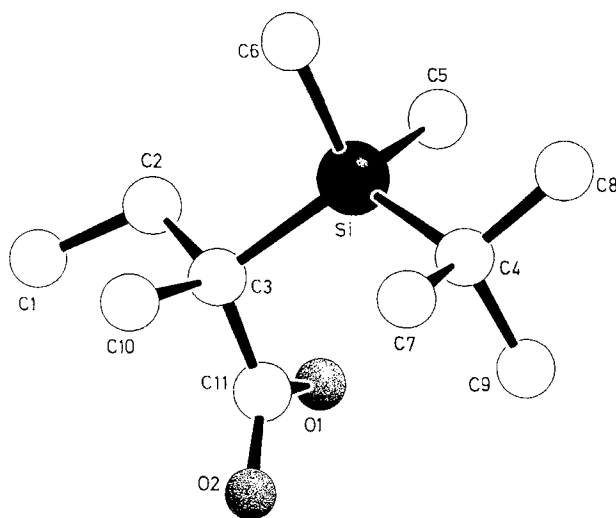


Figure 2. Molecular structure of (*S*)-**16g**^[9] in the solid state (SCHAKAL^[12] plot)

philic substitution to give a quaternary carbon center by using SAMP-/RAMP-hydrazone has been proven. The use of these building blocks for the enantioselective synthesis of β,β -disubstituted γ -lactones and γ -quaternary oxiranes has already been investigated^[4e]. Further applications to the enantioselective synthesis of biologically active compounds and natural products can now be studied.

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Experimental

¹H and ¹³C NMR: in CDCl₃, TMS as internal reference, Varian XL 100, Varian EM-390, Varian VXR 300 or a Unity 500. – IR: Beckman Acculab 4 or Perkin-Elmer Infracord 337. – MS: Kratos MS-30 or Varian MAT 212 (70 eV). – Optical rotation: Room temperature, Perkin Elmer P 241 polarimeter. – Microanalyses: Heraeus CHN-O-RAPID element analyzer. – Ozonolyses: Fischer ozone generator type 502. – TLC analyses: Merck TLC plates silica gel 60 F₂₅₄. – All solvents were dried and distilled according to standard procedures. – Bromoacetaldehyde diethyl acetal was purchased from Janssen, Beerse, Belgium. (*S*)- and (*R*)-1-Amino-2-(methoxymethyl)pyrrolidine (SAMP and RAMP) were synthesized according to the literature procedure^[11a] from (*S*)- and (*R*)-proline.

(*S*)-1-Formyl-2-[[2-(methoxyethoxy)methoxy]methyl]pyrrolidine (**5**): To a solution of 1.29 g (10 mmol) of *N*-formylproline in 20 ml of acetonitrile was added a solution of 3.39 g (15 mmol) of triethyl[[2-(methoxyethoxy)methyl]ammonium chloride^[13] in 25 ml of acetonitrile. After refluxing of the reaction mixture for 14 h and storage at 2°C overnight the precipitate of triethylammonium chloride was filtered, washed with diethyl ether, and the combined organic phases were concentrated in vacuo. Reduced-pressure distillation of the residue afforded 1.86 g (86%) of a colorless oil; b.p. 121–122°C/0.25 Torr, $\alpha_D^{25} = -31.0$ (neat). – IR (neat): $\tilde{\nu} = 2930$ cm⁻¹, 2880, 2810, 1665, 1455, 1410, 1380, 1345, 1240, 1110, 1040, 845, 750, 720. – ¹H NMR (CDCl₃): $\delta = 2.03$ (m, 4H, CH₂pyr), 3.32–3.97 (m, 8H, NCH₂, OCH₂, OCH₂CH₂O), 3.54 (s, 3H, OCH₃), 4.00–4.20 (m, 1H, NCH), 4.84 (s, 2H, OCH₂O), 8.39, 8.46 (s, 1H, CHO, *E/Z* = 3:1). – MS, *m/z* (%): 218 (1) [M⁺ + 1], 217 (2) [M⁺], 112 (73), 111 (53), 98 (100) [C₅H₉NO⁺], 70 (90), 68 (27), 59 (72), 55 (28), 45 (39), 43 (45), 41 (49). – C₁₀H₁₉NO₄ (217.3): calcd. C 55.28, H 8.81, N 6.45; found C 54.85, H 8.81, N 6.70.

(*S*)-2-[[2-(Methoxyethoxy)methoxy]methyl]pyrrolidine (**6**): To a solution of 6.31 g (112.5 mmol) of KOH in 55 ml of water was added slowly 16.30 g (75 mmol) **5**. The resultant mixture was refluxed for 4 h until completion of the reaction (TLC control). After repeated extraction with CH₂Cl₂ the combined organic extracts were dried with Na₂SO₄ and concentrated in vacuo. Reduced-pressure distillation of the residue afforded 13.88 g (98%) of a col-

orless oil; b.p. 78°C/0.15 Torr, $\alpha_D^{25} = -3.5$ (neat). – IR (neat): $\tilde{\nu} = 3350$ (NH) cm⁻¹, 2930, 2875, 1455, 1410, 1365, 1280, 1240, 1195, 1170, 1110, 1045, 980, 930, 845. – ¹H NMR (CDCl₃): $\delta = 1.75$ (m, 4H, CH₂pyr), 2.06 (s, 1H, NH), 2.90 (m, 2H, NCH₂), 3.06–3.82 (m, 7H, OCH₂, NCH₂, OCH₂CH₂O), 3.40 (s, 3H, OCH₃), 4.71 (s, 2H, OCH₂O). – MS, *m/z* (%): 190 (1) [M⁺ + 1], 114 (24) [C₆H₁₂NO⁺], 84 (42) [C₅H₁₀N⁺], 70 (100) [C₄H₈N⁺], 59 (31), 55 (32), 45 (41), 43 (35), 41 (32). – C₉H₁₉NO₂ (189.5): calcd. C 57.11, H 10.21, N 7.40; found C 56.85, H 10.08, N 7.20.

(*S*)-2-[[2-(Methoxyethoxy)methoxy]methyl]-1-nitrosopyrrolidine (**7**): 12.36 g of a 85% solution of ethyl nitrite (140 mmol) was dissolved in 50 ml of THF and the solution cooled to 0°C. To this solution was added dropwise 13.31 g (70.33 mmol) of **6**. The reaction mixture was allowed to warm to room temp. overnight and stirred for 22 h at this temp. (TLC control). After concentration in vacuo the product was distilled under reduced pressure to afford 14.61 g (95%) of a yellow oil; b.p. 118°C/0.1 torr, $\alpha_D^{25} = -65.6$ (neat). – IR (neat): $\tilde{\nu} = 2930$ cm⁻¹, 2880, 2820, 1665, 1450, 1415, 1360, 1300, 1195, 1170, 1150, 1110, 1040, 980, 965, 940, 890, 845, 805, 770. – ¹H NMR (CDCl₃): $\delta = 2.06$ (m, 4H, CH₂pyr), 3.20–3.82 (m, 6H, NCH₂, OCH₂CH₂O), 3.43 (s, 3H, OCH₃), 3.96 (d, 2H, OCH₂), 4.52–4.83 (m, 1H, NCH), 4.74 (s, 2H, OCH₂O). – MS, *m/z* (%): 218 (<1) [M⁺], 89 (27), 70 (37), 59 (100), 55 (39), 45 (63), 42 (37), 41 (55). – C₉H₁₂N₂O₄ (189.5): calcd. C 49.53, H 8.31, N 12.84; found C 49.76, H 8.41, N 12.49.

(*S*)-1-Amino-2-[[2-(methoxyethoxy)methoxy]methyl]pyrrolidine (SAMP)^[4a] (**8**): In a 250-ml three-necked flask 5.28 g (139.1 mmol) of LiAlH₄ was suspended in 120 ml of THF, then 14.02 g (64.24 mmol) of **7** in 30 ml of THF was added dropwise to the suspension during 2 h. The resultant reaction mixture was refluxed for 4 h and stirred overnight. After hydrolysis with 2 ml of a 10% KOH solution and 18 ml of water, the resulting slurry was filtered with suction, and the precipitate was refluxed in THF (150 ml) for 4 h, filtered and then the process repeated with fresh THF (150 ml). The combined THF extracts were concentrated in vacuo. The resultant oil was dissolved in CH₂Cl₂ and the solution dried with Na₂SO₄. After concentration of the solution, reduced-pressure distillation of the residue afforded 11.85 g (90%) of a colorless oil; b.p. 87–88°C/0.1 Torr, $\alpha_D^{25} = -71.5$ (neat). – IR (neat): $\tilde{\nu} = 3350$ (NH₂) cm⁻¹, 3150, 2930, 2880, 2820, 1605, 1460, 1410, 1365, 1280, 1240, 1200, 1170, 1110, 1050, 980, 930, 850. – ¹H NMR (CDCl₃): $\delta = 1.75$ (m, 4H, CH₂pyr), 2.40 (m, 2H, NCH₂), 2.82–3.86 (m, 7H, NCH, OCH₂O, OCH₂CH₂O), 3.18 (s, 2H, NCH₂), 3.42 (s, 3H, OCH₃), 4.76 (s, 2H, OCH₂O). – ¹³C NMR (CDCl₃): $\delta = 20.9$ (NCH₂CH₂), 26.6 (CH₂CH), 58.8 (OCH₃), 60.2 (CH₂N), 66.8 (CH₂OCH₃), 68.3 (CH), 70.3 (CH₂O), 71.9 (OCH₂CH₂), 95.8 (OCH₂O). – MS, *m/z* (%): 205 (2) [M⁺ + 1], 204 (12) [M⁺], 97 (40), 85 (79), 83 (52), 70 (100) [C₄H₈N⁺], 68 (30), 59 (51), 45 (37), 44 (20), 41 (56). – C₉H₂₀N₂O₃ (204.3): calcd. C 52.92, H 9.87, N 13.72; found C 52.90, H 9.94, N 13.36.

(*R,S,S*)-2-(Methoxymethyl)-1-[(2-phenylpropylidene)amino]pyrrolidine (**2a**): To 2.60 g (20 mmol) of SAMP was added dropwise 2.68 g (20 mmol) of (\pm)-2-phenylpropanal at 0°C. After refluxing at 60°C until completion of the reaction (TLC control) the reaction mixture was extracted with CH₂Cl₂/H₂O (4:1). The organic phase was dried with Na₂SO₄ and concentrated in vacuo. Reduced-pressure distillation of the residue afforded 4.30 g (87%) of a light yellow oil; b.p. 112–113°C/0.15 Torr, $\alpha_D^{25} = -91.5$ (neat). – IR (neat): $\tilde{\nu} = 3080$ –2780 cm⁻¹, 1600, 1490, 1450, 1370, 1355, 1300, 1280, 1190, 1140–1080, 1050, 1010, 970, 900, 760, 695. – ¹H NMR (CCl₄): $\delta = 1.35$ (d, 3H, CH₃), 1.80 (m, 4H, NCH₂pyr), 2.65 (m, 1H, NCH₂), 2.95–3.75 (m, 5H, NCH₂, NCH, CH₂O, CH), 3.25 (s, 3H, OCH₃), 6.48 (d, 1H, CHN), 7.15 (m, 5H, C₆H₅). – MS, *m/z* (%): 246 (3) [M⁺], 201 (100) [M⁺ – CH₂OCH₃], 105 (47), 70 (33). – C₁₅H₂₂N₂O (246.4): calcd. C 73.13, H 9.00, N 11.37; found C 73.10, H 8.71, N 11.33.

(*R,S,S*)-2-(Methoxymethyl)-1-[(2-phenylcyclopentylidene)amino]pyrrolidine (**2c**): 7.25 g (89%) of **2c** was prepared from 3.91 g (30 mmol) of SAMP and 4.81 g (30 mmol) of (\pm)-2-phenylcyclopentanone according to the procedure described for the synthesis of **2a**; b.p. 135°C/0.2 Torr, $\alpha_D^{25} = +234.8$ (neat). – IR (neat): $\tilde{\nu} = 3080$ cm⁻¹, 3055, 3020, 2960, 2870, 2820, 2720, 1640 (CN), 1600, 1490, 1445, 1420, 1375, 1335, 1270, 1190, 1100, 1040, 1025, 1000, 960, 905, 840, 750, 695. – ¹H NMR (CDCl₃): $\delta = 1.14$ –2.88 (m, 11H, NCH₂pyr, NCH₂, CH₂), 2.90–3.94 (m, 5H, NCH₂, NCH, CH₂O, CH), 3.33, 3.37 (s, 3H, OCH₃), 7.33 (m, 5H, C₆H₅). – MS, *m/z* (%): 273 (2) [M⁺ + 1], 272 (8) [M⁺], 227 (100) [M⁺ – CH₂OCH₃], 158 (70) [C₁₁H₁₂N⁺], 117 (83), 115 (23), 91 [C₇H₇⁺], 70 (47), 45 (23). – C₁₇H₂₄N₂O (272.4): calcd. C 74.96, H 8.88, N 10.29; found C 75.10, H 8.75, N 10.25.

(*R,S,S*)-2-(Methoxymethyl)-1-[(2-phenylcyclohexylidene)amino]pyrrolidine (**2d**): 6.92 g (81%) of **2d** was prepared from 3.91 g (30 mmol) of SAMP and 5.23 g (30 mmol) of (\pm)-2-phenylcyclohexanone by the same procedure as described for the synthesis of **2a**; b.p. 132–133°C/0.03 Torr, $\alpha_D^{25} = +234.6$ (neat). – IR (neat): $\tilde{\nu} = 3090$ cm⁻¹, 3060, 3030, 2940, 2860, 1635 (CN), 1600, 1495, 1450, 1195, 1125, 1050, 1030, 1000, 970, 920, 845, 770, 725, 700. – ¹H NMR (CDCl₃): $\delta = 1.40$ –2.08 (m, 10H, NCH₂pyr,

C_6H_5). — ^{13}C NMR ($CDCl_3$): δ = 14.0 (CH_3), 22.1 (NCH_2CH_2), 22.5 (CH_2), 25.2 (CH_2), 26.9 (CH_2CH), 29.5 (CH_2), 31.6 (CH_2), 34.8 (CH_2), 48.9 (CH_2N), 52.2 (CO_2CH_3), 59.1 (OCH_3), 61.8 (C), 62.8 (CH), 74.5 (CH_2O), 128.4, 129.1, 131.1 (CH_{arom}), 131.2 (C_{arom}), 137.2 (CHN), 172.1 (CO_2). — MS, *m/z* (%): 406 (1) [M^+], 297 (100) [M^+ — SPh], 114 (19), 70 (52). — $C_{22}H_{34}N_2O_3S$ (406.6): calcd. C 64.99, H 8.43, N 6.89; found C 64.70, H 8.47, N 6.83.

(*S,R*)-(–)-1-[2-(*Methoxycarbonyl*)-2-(phenylthio)decyliden]amino}-2-(methoxymethyl)pyrrolidine (**11e**): 1.13 g (3.0 mmol) of (*S,S*)-2-(methoxymethyl)-1-[2-(phenylthio)decyliden]amino}pyrrolidine^[7] (**10e**) was transformed by treatment with 4.5 mmol of *n*BuLi and 0.48 g (5.1 mmol) of methyl chloroformate into 1.00 g (77%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $\alpha_D^{25} = -19.2$ (neat); de = 92% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3060 cm^{-1} , 3000–2800, 1725 (C=O), 1580 (CN), 1460, 1435, 1340, 1300, 1225, 1195, 1120, 1065, 1020, 750, 705, 690. — 1H NMR ($CDCl_3$): δ = 0.87 (br t, 3H, CH_2CH_3), 1.20–1.52 [m, 12H, (CH_2)₆CH₃], 1.74–1.99 [m, 5H, CH_{2pyr}], 2.16 [m, 1H, $CH_2(CH_2)_6CH_3$], 2.78 (m, 1H, NCH₂), 3.12–3.48 (m, 4H, OCH₂, NCH₂, NCH), 3.33 (s, 3H, OCH₃), 3.69 (s, 3H, CO_2CH_3), 6.48 (s, 1H, CHN), 7.24–7.54 (m, 5H, C_6H_5). — ^{13}C NMR ($CDCl_3$): δ = 14.1 (CH_3), 22.0 (NCH_2CH_2), 22.7 (CH_2), 25.2 (CH_2), 26.8 (CH_2CH), 29.2 (CH_2), 29.3 (CH_2), 29.9 (CH_2), 31.9 (CH_2), 34.7 (CH_2), 48.9 (CH_2N), 52.2 (CO_2CH_3), 59.1 (OCH_3), 61.8 (C), 62.8 (CH), 74.5 (CH_2O), 128.4, 129.1, 131.1 (CH_{arom}), 31.1 (C_{arom}), 137.2 (CHN), 172.1 (CO_2). — MS, *m/z* (%): 434 (1) [M^+], 325 (100) [M^+ — SPh], 114 (25), 70 (58). — $C_{24}H_{38}N_2O_3S$ (434.6): calcd. C 66.32, H 8.81, N 6.45; found C 66.48, H 8.78, N 6.81.

(*S,R*)-(–)-1-[2-(*Methoxycarbonyl*)-2-(phenylthio)dodecyliden]amino}-2-(methoxymethyl)pyrrolidine (**10f**) was transformed by treatment with 4.5 mmol of LDA and 0.48 g (5.1 mmol) of methyl chloroformate into 1.30 g (94%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $\alpha_D^{25} = -14.4$ (neat); de = 91% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3060 cm^{-1} , 3000–2800, 1730 (C=O), 1580 (CN), 1460, 1445, 1435, 1340, 1315, 1295, 1230, 1195, 1125, 1020, 970, 750, 705, 690. — 1H NMR ($CDCl_3$): δ = 0.88 (br t, 3H, CH_2CH_3), 1.22–1.31 [m, 16H, (CH_2)₈CH₃], 1.74–2.00 [m, 5H, CH_{2pyr}], 2.16 [m, 1H, $CH_2(CH_2)_8CH_3$], 2.79 (m, 1H, NCH₂), 3.12–3.48 (m, 4H, OCH₂, NCH₂, NCH), 3.34 (s, 3H, OCH₃), 3.69 (s, 3H, CO_2CH_3), 6.48 (s, 1H, CHN), 7.25–7.55 (m, 5H, C_6H_5). — ^{13}C NMR ($CDCl_3$): δ = 14.1 (CH_3), 22.1 (NCH_2CH_2), 22.7 (CH_2), 25.3 (CH_2), 26.9 (CH_2CH), 29.4 (CH_2), 29.6 (CH_2), 29.63 (CH_2), 29.9 (CH_2), 32.0 (CH_2), 34.8 (CH_2), 49.0 (CH_2N), 52.2 (CO_2CH_3), 59.2 (OCH_3), 61.9 (C), 62.8 (CH), 74.5 (CH_2O), 128.4, 129.1, 131.2 (CH_{arom}), 131.3 (C_{arom}), 137.2 (CHN), 172.2 (CO_2). — MS, *m/z* (%): 462 (<1) [M^+], 353 (100) [M^+ — SPh], 114 (24), 70 (68), 41 (21). — $C_{26}H_{42}N_2O_3S$ (462.7): calcd. C 67.49, H 9.15, N 6.05; found C 67.32, H 9.20, N 6.63.

(*S,R*)-(–)-1-[2-(*Methoxycarbonyl*)-2-(phenylthio)tridecyliden]amino}-2-(methoxymethyl)pyrrolidine (**11g**): 1.25 g (3.0 mmol) of (*S,S*)-2-(methoxymethyl)-1-[2-(phenylthio)tridecyliden]amino}pyrrolidine^[7] (**10g**) was transformed by treatment with 4.5 mmol of *n*BuLi and 0.48 g (5.1 mmol) of methyl chloroformate into 1.30 g (91%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $\alpha_D^{25} = -14.5$ (neat); de = 93% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3060 cm^{-1} , 3000–2800, 1725 (C=O), 1580 (CN), 1460, 1440, 1360, 1240, 1195, 1120, 1025, 910, 750, 730, 705, 690. — 1H NMR ($CDCl_3$): δ = 0.88 (br t, 3H, CH_2CH_3), 1.16–1.52 [m, 18H, (CH_2)₉CH₃], 1.74–2.01 [m, 5H, CH_{2pyr}], 2.17 [m, 1H, $CH_2(CH_2)_9CH_3$], 2.78 (m, 1H, NCH₂), 3.12–3.50 (m, 4H, OCH₂, NCH₂, NCH), 3.33 (s, 3H, OCH₃), 3.68 (s, 3H, CO_2CH_3), 6.48 (s, 1H, CHN), 7.24–7.54 (m, 5H, C_6H_5). — ^{13}C NMR ($CDCl_3$): δ = 14.1 (CH_3), 22.1 (NCH_2CH_2), 22.7 (CH_2), 25.3 (CH_2), 26.9 (CH_2CH), 29.4 (CH_2), 29.6 (CH_2), 29.7 (CH_2), 29.9 (CH_2), 32.0 (CH_2), 34.8 (CH_2), 48.9 (CH_2N), 52.2 (CO_2CH_3), 59.1 (OCH_3), 61.8 (C), 62.8 (CH), 74.5 (CH_2O), 128.4, 129.0, 131.1 (CH_{arom}), 131.3 (C_{arom}), 137.2 (CHN), 172.1 (CO_2). — MS, *m/z* (%): 476 (2) [M^+], 415 (22), 367 (100) [M^+ — SPh], 269 (33), 168 (28), 114 (32), 110 (57), 109 (30), 77 (20), 70 (93), 55 (52), 43 (44), 41 (73). — $C_{27}H_{44}N_2O_3S$ (476.7): calcd. C 68.03, H 9.39, N 5.88; found C 67.89, H 9.41, N 6.24.

(*S,R*)-(–)-1-[2-(*Methoxycarbonyl*)-2-(phenylthio)-3-(3,4,5-trimethoxyphenyl)propyliden]amino}-2-(methoxymethyl)pyrrolidine (**11h**): 2.66 g (6.0 mmol) of (*S,S*)-2-(methoxymethyl)-1-[2-(phenylthio)-3-(3,4,5-trimethoxyphenyl)propyliden]amino}pyrrolidine^[21] (**10h**) dissolved in 2 ml of THF was transformed by treatment with 9 mmol of *n*BuLi and 0.96 g (10.2 mmol) of methyl chloroformate into 1.80 g (60%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $[\alpha]_D^{25} = -26.3$ (c = 1.14, C_6H_6); de = 93% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3040 cm^{-1} , 2920, 2860, 2815, 1720 (C=O), 1580 (CN), 1495, 1450, 1320, 1230, 1190, 1105, 1040, 1000, 965, 900, 830, 770, 740, 720, 685. — 1H NMR ($CDCl_3$): δ = 1.73–2.00 (m, 4H, CH_{2pyr}), 2.79 (m, 1H, NCH₂), 3.10–3.52 (m, 4H, OCH₂, NCH₂, NCH), 3.35 (s, 3H, CH_2OCH_3), 3.66 (s, 3H, CO_2CH_3), 3.76 (s, 6H, OCH₃), 3.79 (s, 3H, OCH₃), 6.43–6.52 (m, 3H, CHN, C_6H_5), 7.26–7.52 (m, 5H, C_6H_5). — ^{13}C NMR ($CDCl_3$): δ = 22.1 (NCH_2CH_2), 27.2 (CH_2CH), 41.6 (CH_2), 49.0 (CH_2N),

52.2 (CO_2CH_3), 55.9 (OCH_3), 59.2 (OCH_3), 60.8 (OCH_3), 61.9 (C), 62.7 (CH), 74.6 (CH_2O), 108.0, 128.4, 129.0, 131.1 (CH_{arom}), 131.3 (C_{arom}), 132.7 (C_{arom}), 136.6 (C_{arom}), 137.4 (CHN), 152.4 (C_{arom}), 171.5 (CO_2). — MS, *m/z* (%): 457 (1) [M^+ — CH_2OCH_3], 394 (23) [M^+ — CO_2CH_3], 393 (100) [M^+ — SPh], 361 (38), 278 (25), 181 (34) [(H_3CO)₃ $C_6H_2CH_2^+$], 129 (20), 114 (13) [$NC_6H_4O^+$], 70 (37) [$NC_4H_8^+$]. — $C_{26}H_{36}N_2O_6S$ (502.6): calcd. C 62.13, H 6.82, N 5.57; found C 61.05, H 6.90, N 5.86.

(*S,R*)-(–)-1-[2-(*Methoxycarbonyl*)-2-(methylthio)butyliden]amino}-2-(methoxymethyl)pyrrolidine (**11i**): 0.345 g (1.5 mmol) of (*S,R*)-2-(methoxymethyl)-1-[2-(methylthio)butyliden]amino}pyrrolidine^[22] (**10i**) was transformed by treatment with 2.25 mmol of MeLi and 0.24 g (2.55 mmol) of methyl chloroformate into 0.353 g (82%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $\alpha_D^{25} = -85.0$ (neat); de = 88% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3020–2800 cm^{-1} , 1740, 1585, 1465, 1440, 1350, 1240, 1120, 1025. — 1H NMR ($CDCl_3$): δ = 0.95 (t, J = 7.3 Hz, 3H, CH_2CH_3), 1.75–2.25 (m, 6H, CH_{2pyr}), 2.05 (s, 3H, SCH₃), 2.85 (m, 1H, NCH₂), 3.30–3.60 (m, 4H, OCH₂, NCH₂, NCH), 3.35 (s, 3H, OCH₃), 3.76 (s, 3H, CO_2CH_3), 6.55 (s, 1H, CHN). — ^{13}C NMR ($CDCl_3$): δ = 9.5 (CH_3), 12.3 (SCH₃), 22.1 (NCH_2CH_2), 26.6 (CH_2CH), 27.1 (CH_2), 49.2 (CH_2N), 52.4 (CO_2CH_3), 58.6 (C), 59.2 (OCH_3), 63.1 (CH), 7.74 (CH_2O), 131.8 (CHN), 172.2 (CO_2). — MS, *m/z* (%): 288 (4) [M^+], 243 (33), 241 (79), 195 (27), 114 (21), 87 (32), 70 (100). — $C_{13}H_{24}N_2O_3S$ (288.4): calcd. C 54.21, H 8.39, N 9.73; found C 54.04, H 8.50, N 9.85.

(*S,S*)-(–)-1-[2-(*Methoxycarbonyl*)methyl]-2-(methylthio)butyliden]amino}-2-(methoxymethyl)pyrrolidine (**11j**): 0.345 g (1.5 mmol) of (**10j**)^[22] was transformed by treatment with 2.25 mmol of MeLi and 0.36 g (2.55 mmol) of methyl bromoacetate into 0.119 g (26%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $[\alpha]_D^{25} = -79.3$ (c = 1.13, C_6H_6); de = 91% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3020–2800 cm^{-1} , 1745, 1595, 1450, 1350, 1300, 1250, 1100, 1025, 950. — 1H NMR ($CDCl_3$): δ = 1.00 (t, J = 7.4 Hz, 3H, CH_2CH_3), 1.75–2.05 (m, 6H, CH_{2pyr}), 1.95 (s, 3H, SCH₃), 2.77 (m, 3H, CH_2CO_2 , NCH₂), 3.30–3.60 (m, 4H, OCH₂, NCH₂, NCH), 3.37 (s, 3H, OCH₃), 3.68 (s, 3H, CO_2CH_3), 6.43 (s, 1H, CHN). — ^{13}C NMR ($CDCl_3$): δ = 8.7 (CH_3), 11.0 (SCH₃), 21.9 (NCH_2CH_2), 26.5 (CH_2CH), 28.5 (CH_2), 37.4 (CH_2CO_2), 49.7 (CH_2N), 51.3 (CO_2CH_3), 51.9 (C), 59.2 (OCH_3), 63.7 (CH), 74.4 (CH_2O), 138.2 (CHN), 171.3 (CO_2). — MS, *m/z* (%): 302 (1) [M^+], 255 (38) [M^+ — SCH₃], 209 (100), 108 (25), 70 (52). — $C_{14}H_{26}N_2O_3S$ (302.4): calcd. C 55.59, H 8.67, N 9.26; found C 55.36, H 8.54, N 9.16.

(*S,S*)-(–)-1-[2-(*Methoxycarbonyl*)methyl]-2-(phenylthio)butyliden]amino}-2-(methoxymethyl)pyrrolidine (**11k**): 0.88 g (3.0 mmol) of (*S,S*)-**10a**^[7] was transformed by treatment with 4.5 mmol of MeLi and 0.78 g (5.1 mmol) of methyl bromoacetate into 0.82 g (75%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $[\alpha]_D^{25} = -31.1$ (c = 0.97, C_6H_6); de = 83% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3075 cm^{-1} , 3060, 2960–2830, 1740 (CO), 1585, 1475, 1460, 1440, 1340, 1290, 1200, 1175, 1125, 750, 695. — 1H NMR ($CDCl_3$): δ = 1.06 (t, J = 7.4 Hz, 3H, CH_2CH_3), 1.64–2.12 (m, 6H, CH_{2pyr}), 2.64 (m, 1H, NCH₂), 2.78 (m, 3H, CH_2CO_2), 2.94–3.04 (m, 1H, NCH₂), 3.20–3.28 (m, 3H, OCH₂, NCH), 3.28 (s, 3H, OCH₃), 3.67 (s, 3H, CO_2CH_3), 6.43 (s, 1H, CHN), 7.22–7.47 (m, 5H, C_6H_5). — ^{13}C NMR ($CDCl_3$): δ = 9.0 (CH_3), 22.1 (NCH_2CH_2), 26.6 (CH_2CH), 29.2 (CH_2), 38.5 (CH_2CO_2), 49.2 (CH_2N), 51.2 (CO_2CH_3), 56.8 (C), 59.1 (OCH_3), 63.1 (CH), 74.4 (CH_2O), 128.3, 128.8, 137.2 (CH_{arom}), 131.3 (C_{arom}), 137.7 (CHN), 171.3 (CO_2). — MS, *m/z* (%): 333 (2) [M^+ — OCH₃], 255 (100) [M^+ — SPh], 223 (26), 209 (50), 114 (22) [$NC_6H_4O^+$], 110 (46), 109 (15) [$SC_6H_5^+$], 70 (45) [$NC_4H_8^+$]. — $C_{19}H_{26}N_2O_3S$ (364.5): calcd. C 62.60, H 7.74, N 7.69; found C 62.67, H 7.72, N 7.60.

(*S,S*)-(–)-1-[2-(*Methoxycarbonyl*)methyl]-2-(phenylthio)octyliden]amino}-2-(methoxymethyl)pyrrolidine (**11l**): 1.03 g (3.0 mmol) of (*S,S*)-**10d**^[7] was transformed by treatment with 4.5 mmol of MeLi and 0.73 g (4.8 mmol) of methyl bromoacetate into 0.80 g (64%) of a yellow oil by the procedure described for (*S,R*)-**11a**; $[\alpha]_D^{25} = -15.9$ (c = 1.06, C_6H_6); de = 87% (^{13}C NMR). — IR (neat): $\tilde{\nu}$ = 3060 cm^{-1} , 2950–2850, 1740 (CO), 1590, 1455, 1435, 1340, 1300, 1280, 1200, 1170, 1120, 1020, 970, 910, 750, 730. — 1H NMR ($CDCl_3$): δ = 0.88 (t, J = 6.8 Hz, 3H, CH_2CH_3), 1.27–2.08 [m, 14H, CH_{2pyr}], 2.58–2.67 (m, 1H, NCH₂), 2.77 (d, J = 1.7 Hz, 3H, CH_2CO_2), 2.90–3.01 (m, 1H, NCH₂), 3.20–3.37 (m, 3H, OCH₂, NCH), 3.27 (s, 3H, OCH₃), 3.66 (s, 3H, CO_2CH_3), 6.42 (s, 1H, CHN), 7.21–7.45 (m, 5H, C_6H_5). — ^{13}C NMR ($CDCl_3$): δ = 14.1 (CH_3), 22.1 (NCH_2CH_2), 22.7 (CH_2), 24.4 (CH_2), 26.5 (CH_2CH), 29.7 (CH_2), 31.7 (CH_2), 36.5 (CH_2), 38.9 (CH_2CO_2), 49.1 (CH_2N), 51.1 (CO_2CH_3), 56.4 (C), 59.1 (OCH_3), 63.1 (CH), 74.4 (CH_2O), 128.2, 128.8, 137.3 (CH_{arom}), 131.4 (C_{arom}), 137.7 (CHN), 171.3 (CO_2). — MS, *m/z* (%): 389 (1) [M^+ — OCH₃], 311 (100) [M^+ — SPh], 265 (52), 114 (15) [$NC_6H_4O^+$], 110 (52), 109 (15) [$SC_6H_5^+$], 70 (39) [$NC_4H_8^+$]. — $C_{23}H_{36}N_2O_3S$ (420.6): calcd. C 65.58, H 8.63, N 6.66; found C 65.85, H 8.64, N 6.56.

(*R*)-(+)-Methyl 2-Formyl-2-(phenylthio)butanoate (**12a**): 0.70 g (2.0 mmol) of (*S,R*)-**11a** was cleaved with 18 ml of 5 N HCl (18.5 h) by the

[M⁺], 379 (100) [M⁺ - CH₂OCH₃], 310 (28), 115 (13) [SiC₆H₁₃⁺], 114 (28) [NC₆H₁₂O⁺], 73 (61). - C₂₆H₅₄N₂O₃Si (424.8); calcd. C 70.69, H 12.34, N 6.60; found C 70.71, H 12.30, N 7.18.

(*S,S*)-(-)-1-[[2-(*tert*-Butyldimethylsilyl)tetradecyliden]amino]-2-(methoxymethyl)pyrrolidine (**14h**): 3.21 g (11.9 mmol) of (*S,S*)-**13a**⁶ was transformed by treatment with 17.9 mmol of LDA in THF and 4.46 g (17.9 mmol) of 1-bromododecane into 3.39 g (65%) of a yellow oil by the procedure described for (*S,S*)-**14a**; [α]_D²⁵ = -62.5 (c = 1.12, C₆H₆); de = 96% (¹³C NMR). - IR (neat): ν = 2960–2850 cm⁻¹, 1460, 1365, 1340, 1245, 1195, 1120, 1010, 970, 905, 830, 805, 770. - ¹H NMR (CDCl₃): δ = 0.02 [s, 3H, Si(CH₃)₂], 0.03 [s, 3H, Si(CH₃)₂], 0.92 (t, J = 7.1 Hz, 3H, CH₃), 0.95 [s, 9H, SiC(CH₃)₃], 1.17–2.06 (m, 27H, CH₂, CH, CH₂pyr), 2.76 (m, 1H, NCH₂), 3.09–3.58 (m, 4H, NCH, NCH₂, OCH₂), 3.38 (s, 3H, OCH₃), 6.67 (d, J = 8.4 Hz, 1H, CHN). - ¹³C NMR (CDCl₃): δ = -6.8 [Si(CH₃)₂], -6.7 [Si(CH₃)₂], 14.1 (CH₃), 17.6 (C), 21.9 (NCH₂CH₂), 22.7 (CH₂), 26.4 (NCHCH₂), 27.3 [SiC(CH₃)₃], 28.4 (CH₂), 29.4 (CH₂), 29.58 (CH₂), 29.60 (CH₂), 29.67 (CH₂), 29.73 (CH₂), 29.80 (CH₂), 29.81 (CH₂), 30.5 (CH₂), 32.0 (CH₂), 32.3 (CH), 51.1 (NCH₂), 59.1 (OCH₃), 63.9 (NCH), 74.9 (CH₂O), 144.4 (CHN). - MS, m/z (%): 439 (7) [M⁺ + 1], 438 (18) [M⁺], 393 (100) [M⁺ - CH₂OCH₃], 324 (32), 115 (15) [SiC₆H₁₃⁺], 114 (34) [NC₆H₁₂O⁺], 73 (73). - C₂₆H₅₄N₂O₃Si (424.8); calcd. C 71.17, H 12.40, N 6.38; found C 70.76, H 12.37, N 6.25.

(*S,R*)-(+)-1-[[2-(*tert*-Butyldimethylsilyl)-2-(methoxycarbonyl)-butyliden]amino]-2-(methoxymethyl)pyrrolidine (**15a**): 3.37 g (11.3 mmol) of (*S,S*)-**14b** was added at -110°C to a solution of 50 mmol of MeLi in 35 ml of THF. After stirring for 1 h at this temp. the reaction mixture was allowed to warm to room temp. overnight. The metalated hydrazone was cooled to -110°C, and 7.56 g (80 mmol) of methyl chloroformate was added. After stirring for 3 h at this temp. the reaction mixture was allowed to warm to room temp. overnight. It was subsequently diluted with diethyl ether, washed with pH 7 buffer, then with water and dried with MgSO₄. Flash chromatography yielded 2.27 g (56%) of a yellow oil; [α]_D²⁵ = +7.1 (c = 1.48, C₆H₆); de = 90% (¹³C NMR). - IR (neat): ν = 3000–2800 cm⁻¹, 1705, 1590, 1460, 1450, 1370, 1320, 1295, 1220, 1130, 1110, 1020, 970, 825, 820, 770, 690, 665. - ¹H NMR (CDCl₃): δ = 0.03 [s, 3H, Si(CH₃)₂], 0.11 [s, 3H, Si(CH₃)₂], 0.83 (t, J = 7.1 Hz, 3H, CH₃), 0.89 [s, 9H, SiC(CH₃)₃], 1.73–2.22 (m, 6H, CH₂, CH₂pyr), 2.76 (m, 1H, NCH₂), 3.28–3.65 (m, 4H, NCH, NCH₂, OCH₂), 3.37 (s, 3H, OCH₃), 3.69 (s, 3H, CO₂CH₃), 7.14 (s, 1H, CHN). - ¹³C NMR (CDCl₃): δ = -6.7 [Si(CH₃)₂], -6.5 [Si(CH₃)₂], 10.2 (CH₃), 19.3 [SiC(CH₃)₃], 21.9 (NCH₂CH₂), 23.7 (CH₂), 26.5 (CH₂CH), 27.6 [SiC(CH₃)₃], 49.7 (NCH₂, C), 51.1 (CO₂CH₃), 59.2 (OCH₃), 63.6 (NCH), 74.9 (CH₂O), 147.4 (CHN), 175.1 (CO₂). - MS, m/z (%): 357 (16) [M⁺ + 1], 356 (63) [M⁺], 311 (100) [M⁺ - CH₂OCH₃], 115 (15) [SiC₆H₁₃⁺], 73 (63). - C₁₈H₃₆N₂O₃Si (356.6); calcd. C 60.63, H 10.18, N 7.86; found C 60.59, H 10.28, N 7.63.

(*S,R*)-(-)-1-[[2-(*tert*-Butyldimethylsilyl)-2-(methoxycarbonyl)-pentyliden]amino]-2-(methoxymethyl)pyrrolidine (**15b**): 0.74 g (2.4 mmol) of (*S,S*)-**14c** was transformed by treatment with 11.8 mmol of MeLi and 1.19 g (18.9 mmol) of methyl chloroformate to 0.42 g (47%) of a yellow oil by the procedure described for (*S,R*)-**15a**; [α]_D²⁵ = -12.0 (neat), de value = 96% (¹³C NMR). - IR (neat): ν = 3000–2800 cm⁻¹, 1710, 1590, 1460, 1435, 1390, 1360, 1340, 1295, 1250, 1210, 1110, 1020, 970, 905, 835, 825, 805, 775, 690, 665. - ¹H NMR (CDCl₃): δ = 0.03 [s, 3H, Si(CH₃)₂], 0.12 [s, 3H, Si(CH₃)₂], 0.87 (t, J = 7.4 Hz, 3H, CH₃), 0.89 [s, 9H, SiC(CH₃)₃], 1.00–1.52 (m, 2H, CH₂), 1.73–2.08 (m, 6H, CH₂, CH₂pyr), 2.78 (m, 1H, NCH₂), 3.27–3.62 (m, 4H, NCH, NCH₂, OCH₂), 3.36 (s, 3H, OCH₃), 3.68 (s, 3H, CO₂CH₃), 7.16 (s, 1H, CHN). - ¹³C NMR (CDCl₃): δ = -6.6 [Si(CH₃)₂], -6.5 [Si(CH₃)₂], 14.7 (CH₃), 19.0 (CH₃), 19.3 [SiC(CH₃)₃], 21.9 (NCH₂CH₂), 26.4 (CH₂CH), 27.6 [SiC(CH₃)₃], 33.0 (CH₂), 33.0 (CH₂), 49.2 (C), 49.6 (NCH₂), 51.1 (CO₂CH₃), 59.2 (OCH₃), 63.7 (NCH), 74.7 (CH₂O), 137.7 (CHN), 175.2 (CO₂). - MS, m/z (%): 370 (12) [M⁺], 325 (35) [M⁺ - CH₂OCH₃], 115 (25) [SiC₆H₁₃⁺], 114 (10) [NC₆H₁₂O⁺], 89 (28), 73 (100), 70 (40) [NC₆H₁₂O⁺]. - C₁₉H₃₈N₂O₃Si (370.6); calcd. C 61.58, H 10.33, N 7.56; found C 61.60, H 10.56, N 7.80.

(*S,R*)-(+)-1-[[2-(*tert*-Butyldimethylsilyl)-2-(methoxycarbonyl)-hexyliden]amino]-2-(methoxymethyl)pyrrolidine (**15c**): 0.70 g (2.15 mmol) of (*S,S*)-**14d** was transformed with 10.7 mmol of MeLi and 1.62 g (17.2 mmol) of methyl chloroformate into 0.51 g (62%) of a yellow oil by the procedure described for (*S,R*)-**15a**; [α]_D²⁵ = +7.1 (c = 0.98, C₆H₆); de = 97% (¹³C NMR). - IR (neat): ν = 3000–2800 cm⁻¹, 1710, 1590, 1460, 1430, 1390, 1380, 1360, 1340, 1240, 1200, 1110, 1030, 1010, 855, 835, 800, 770, 690, 665. - ¹H NMR (CDCl₃): δ = 0.03 [s, 3H, Si(CH₃)₂], 0.12 [s, 3H, Si(CH₃)₂], 0.88 (t, J = 7.4 Hz, 3H, CH₃), 0.89 [s, 9H, SiC(CH₃)₃], 0.87–1.50 (m, 4H, CH₂), 1.76–2.11 (m, 6H, CH₂, CH₂pyr), 2.76 (m, 1H, NCH₂), 3.29–3.62 (m, 4H, NCH, NCH₂, OCH₂), 3.36 (s, 3H, OCH₃), 3.68 (s, 3H, CO₂CH₃), 7.16 (s, 1H, CHN). - ¹³C NMR (CDCl₃): δ = -6.6 [Si(CH₃)₂], -6.5 [Si(CH₃)₂], 14.0 (CH₃), 19.3 [SiC(CH₃)₃], 21.8 (NCH₂CH₂), 21.9 (CH₂), 26.4 (CH₂CH), 27.6 [SiC(CH₃)₃], 27.9 (CH₂), 30.5 (CH₂), 49.1 (C), 49.6 (NCH₂),

51.1 (CO₂CH₃), 59.2 (OCH₃), 63.7 (NCH), 74.7 (CH₂O), 137.7 (CHN), 175.2 (CO₂). - MS, m/z (%): 385 (13) [M⁺ + 1], 384 (43) [M⁺], 339 (100) [M⁺ - CH₂OCH₃], 115 (24) [SiC₆H₁₃⁺], 114 (11) [NC₆H₁₂O⁺], 73 (86), 70 (35) [NC₆H₈⁺]. - C₂₀H₄₀N₂O₃Si (384.6); calcd. C 62.45, H 10.48, N 7.28; found C 62.43, H 10.42, N 7.38.

(*S,R*)-1-[[2-(*tert*-Butyldimethylsilyl)-2-(methoxycarbonyl)octyliden]amino]-2-(methoxymethyl)pyrrolidine (**15d**): 3.55 g (10 mmol) of (*S,S*)-**14e** was transformed by treatment with 50 mmol of MeLi and 7.56 g (80 mmol) of methyl chloroformate into 2.15 g (52%) of a yellow oil by the procedure described for (*S,R*)-**15a**; [α]_D²⁵ = 0.0 (c = 0.95, C₆H₆); de = 98% (¹³C NMR). - IR (neat): ν = 3000–2780 cm⁻¹, 1710, 1460, 1390, 1380, 1360, 1340, 1220, 1190, 1110, 1040, 1005, 970, 860, 835, 805, 770, 690, 665. - ¹H NMR (CDCl₃): δ = 0.03 [s, 3H, Si(CH₃)₂], 0.12 [s, 3H, Si(CH₃)₂], 0.87 (t, J = 7.1 Hz, 3H, CH₃), 0.89 [s, 9H, SiC(CH₃)₃], 1.21–1.34 (m, 8H, CH₂), 1.79–2.10 (m, 6H, CH₂, CH₂pyr), 2.77 (m, 1H, NCH₂), 3.27–3.63 (m, 4H, NCH, NCH₂, OCH₂), 3.37 (s, 3H, OCH₃), 3.68 (s, 3H, CO₂CH₃), 7.15 (s, 1H, CHN). - ¹³C NMR (CDCl₃): δ = -6.6 [Si(CH₃)₂], -6.5 [Si(CH₃)₂], 14.1 (CH₃), 19.3 [SiC(CH₃)₃], 21.9 (NCH₂CH₂), 22.7 (CH₂), 25.6 (CH₂), 26.5 (CH₂CH), 27.6 [SiC(CH₃)₃], 30.0 (CH₂), 30.8 (CH₂), 31.8 (CH₂), 49.2 (C), 49.6 (NCH₂), 51.1 (CO₂CH₃), 59.2 (OCH₃), 63.7 (NCH), 74.8 (CH₂O), 137.8 (CHN), 175.2 (CO₂). - MS, m/z (%): 413 (19) [M⁺ + 1], 412 (63) [M⁺], 367 (100) [M⁺ - CH₂OCH₃], 115 (15) [SiC₆H₁₃⁺], 114 (9) [NC₆H₁₂O⁺], 73 (55). - C₂₂H₄₄N₂O₃Si (412.7); calcd. C 64.03, H 10.75, N 6.79; found C 64.18, H 10.66, N 7.22.

(*S,R*)-(-)-1-[[2-(*tert*-Butyldimethylsilyl)-2-(methoxycarbonyl)decyldien]amino]-2-(methoxymethyl)pyrrolidine (**15e**): 1.91 g (5.0 mmol) of (*S,S*)-**14f** was transformed by treatment with 25 mmol of MeLi and 3.78 g (40 mmol) of methyl chloroformate into 0.95 g (43%) of a yellow oil by the procedure described for (*S,R*)-**15a**; [α]_D²⁵ = +4.3 (c = 1.20, C₆H₆); de = 96% (¹³C NMR). - IR (neat): ν = 3000–2800 cm⁻¹, 1710, 1460, 1390, 1380, 1360, 1340, 1250, 1220, 1170, 1115, 835, 820, 770. - ¹H NMR (CDCl₃): δ = 0.02 [s, 3H, Si(CH₃)₂], 0.11 [s, 3H, Si(CH₃)₂], 0.87 (t, J = 7.1 Hz, 3H, CH₃), 0.89 [s, 9H, SiC(CH₃)₃], 1.25 (m, 12H, CH₂), 1.75–2.09 (m, 6H, CH₂, CH₂pyr), 2.76 (m, 1H, NCH₂), 3.29–3.63 (m, 4H, NCH, NCH₂, OCH₂), 3.37 (s, 3H, OCH₃), 3.68 (s, 3H, CO₂CH₃), 7.15 (s, 1H, CHN). - ¹³C NMR (CDCl₃): δ = -6.6 [Si(CH₃)₂], -6.5 [Si(CH₃)₂], 14.1 (CH₃), 19.3 [SiC(CH₃)₃], 21.9 (NCH₂CH₂), 22.7 (CH₂), 25.7 (CH₂), 26.5 (CH₂CH), 27.6 [SiC(CH₃)₃], 29.4 (CH₂), 29.6 (CH₂), 30.4 (CH₂), 30.8 (CH₂), 31.9 (CH₂), 49.1 (C), 49.6 (NCH₂), 51.1 (CO₂CH₃), 59.2 (OCH₃), 63.7 (NCH), 74.8 (CH₂O), 137.8 (CHN), 175.2 (CO₂). - MS, m/z (%): 441 (19) [M⁺ + 1], 440 (59) [M⁺], 395 (100) [M⁺ - CH₂OCH₃], 115 (17) [SiC₆H₁₃⁺], 114 (10) [NC₆H₁₂O⁺], 73 (65). - C₂₄H₄₈N₂O₃Si (440.7); calcd. C 65.41, H 10.98, N 6.36; found C 65.48, H 11.00, N 6.70.

(*R,R*)-(-)-1-[[2-(*tert*-Butyldimethylsilyl)-2-(methoxycarbonyl)decyldien]amino]-2-(methoxymethyl)pyrrolidine (**15e**): 0.76 g (2.0 mmol) of (*R,R*)-**14f** was transformed by treatment with 10 mmol of MeLi and 1.51 g (16 mmol) of methyl chloroformate into 0.28 g (32%) of a yellow oil by the procedure described for (*S,R*)-**15a**; [α]_D²⁵ = -5.6 (c = 1.18, C₆H₆); de = 98% (¹³C NMR). The spectroscopic data were in agreement with those described for (*S,R*)-**15e**.

(*S,R*)-(+)-1-[[2-(*tert*-Butyldimethylsilyl)-2-(methoxycarbonyl)tridecyldien]amino]-2-(methoxymethyl)pyrrolidine (**15f**): 0.85 g (2.0 mmol) of (*S,S*)-**14g** was transformed by treatment with 10 mmol of MeLi and 1.51 g (16 mmol) of methyl chloroformate into 0.60 g (62%) of a yellow oil by the procedure described for (*S,R*)-**15a**; [α]_D²⁵ = +4.1 (c = 1.02, C₆H₆); de = 96% (¹³C NMR). - IR (neat): ν = 3000–2800 cm⁻¹, 1710, 1460, 1390, 1380, 1360, 1340, 1250, 1220, 1115, 860, 835, 805, 770. - ¹H NMR (CDCl₃): δ = 0.02 [s, 3H, Si(CH₃)₂], 0.11 [s, 3H, Si(CH₃)₂], 0.88 (t, J = 7.1 Hz, 3H, CH₃), 0.89 [s, 9H, SiC(CH₃)₃], 1.25 (m, 18H, CH₂), 1.75–2.13 (m, 6H, CH₂, CH₂pyr), 2.76 (m, 1H, NCH₂), 3.29–3.62 (m, 4H, NCH, NCH₂, OCH₂), 3.36 (s, 3H, OCH₃), 3.68 (s, 3H, CO₂CH₃), 7.15 (s, 1H, CHN). - ¹³C NMR (CDCl₃): δ = -6.7 [Si(CH₃)₂], -6.5 [Si(CH₃)₂], 14.1 (CH₃), 19.3 [SiC(CH₃)₃], 21.8 (NCH₂CH₂), 22.6 (CH₂), 25.7 (CH₂), 26.4 (CH₂CH), 27.6 [SiC(CH₃)₃], 29.4 (CH₂), 29.6 (CH₂), 29.67 (CH₂), 29.70 (CH₂), 29.74 (CH₂), 30.4 (CH₂), 30.8 (CH₂), 31.9 (CH₂), 49.1 (C), 49.6 (NCH₂), 51.1 (CO₂CH₃), 59.2 (OCH₃), 63.7 (NCH), 74.8 (CH₂O), 137.7 (CHN), 175.2 (CO₂). - MS, m/z (%): 483 (25) [M⁺ + 1], 482 (70) [M⁺], 437 (100) [M⁺ - CH₂OCH₃], 115 (15) [SiC₆H₁₃⁺], 114 (11) [NC₆H₁₂O⁺], 73 (63). - C₂₇H₅₄N₂O₃Si (482.8); calcd. C 67.17, H 11.27, N 5.80; found C 67.26, H 11.27, N 6.35.

(*S,S*)-(-)-1-[[2-(*tert*-Butyldimethylsilyl)-2-methylbutyliden]amino]-2-(methoxymethyl)pyrrolidine (**15g**): 2.84 g (10 mmol) of (*S,S*)-**14b** was metalated with 50 mmol of MeLi and alkylated with 12.50 g (80 mmol) of iodoethane by the procedure described for (*S,R*)-**15a**. The reaction mixture was diluted with diethyl ether, washed with brine, then with water and dried with MgSO₄. Flash chromatography yielded 2.57 g (82%) of a yellow oil; [α]_D²⁵ = -120.5 (c = 1.00, C₆H₆); de = 93% (¹³C NMR). - IR (neat): ν = 2960–2820 cm⁻¹, 1460, 1380, 1360, 1340, 1250, 1200, 1120, 1010, 970, 830,

(*R*)-(-)-Methyl 2-(*tert*-Butyldimethylsilyl)-2-formylpentanoate (**16b**): 0.37 g (1.0 mmol) of (*S,R*)-**15b** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.17 g (66%) of a colorless liquid; $[\alpha]_D^{25} = -167.9$ ($c = 0.97$, C_6H_6); ee = 96% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 2950$ cm^{-1} , 2920, 2850, 2740, 1700, 1460, 1430, 1260, 1200, 1110, 930, 905, 860, 840, 820. - 1H NMR (CDCl₃): $\delta = 0.09$ [s, 3H, Si(CH₃)₂], 0.17 [s, 3H, Si(CH₃)₂], 0.87 (t, $J = 7.4$ Hz, 3H, CH), 0.92 [d, $J = 0.7$ Hz, 9H, SiC(CH₃)₃], 1.06-1.31 (m, 2H, CH₂), 1.77-1.88 (m, 1H, CH₂), 1.99-2.10 (m, 1H, CH₂), 3.78 (d, $J = 0.7$ Hz, 3H, OCH₃), 10.22 (d, $J = 1.3$ Hz, 1H, CHO). - ^{13}C NMR (CDCl₃): $\delta = -7.0$ [Si(CH₃)₂], -6.7 [Si(CH₃)₂], 14.5 (CH₃), 19.3 [SiC(CH₃)₃], 20.0 (CH₂), 27.1 [SiC(CH₃)₃], 31.1 (CH₂), 51.5 (OCH₃), 63.3 (C), 172.5 (CO₂), 201.0 (CHO). - MS, m/z (%): 257 (2) [M⁺ - 1], 201 (61), 89 (100), 73 (33), 59 (25) [CH₃CO₂⁺], 44 (19). - C₁₃H₂₆O₃Si (258.4): calcd. C 60.42, H 10.14; found C 59.93, H 10.31.

(*R*)-(-)-Methyl 2-(*tert*-Butyldimethylsilyl)-2-formylhexanoate (**16c**): 0.35 g (0.90 mmol) of (*S,R*)-**15c** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.20 g (82%) of a colorless liquid; $[\alpha]_D^{25} = -124.9$ ($c = 1.18$, C_6H_6); ee = 97% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 2960$ -2850 cm^{-1} , 2745, 1705, 1465, 1430, 1365, 1245, 1205, 1115, 900, 840, 820, 770, 675. - 1H NMR (CDCl₃): $\delta = 0.09$ [s, 3H, Si(CH₃)₂], 0.17 [s, 3H, Si(CH₃)₂], 0.86 (t, $J = 7.1$ Hz, 3H, CH₃), 0.92 [s, 9H, SiC(CH₃)₃], 1.01-1.40 (m, 4H, CH₂), 1.80-1.91 (m, 1H, CH₂), 2.02-2.13 (m, 1H, CH₂), 3.79 (s, 3H, OCH₃), 10.22 (d, $J = 1.3$ Hz, 1H, CHO). - ^{13}C NMR (CDCl₃): $\delta = -7.0$ [Si(CH₃)₂], -6.8 [Si(CH₃)₂], 13.8 (CH₃), 19.3 [SiC(CH₃)₃], 23.2 (CH₂), 27.1 [SiC(CH₃)₃], 28.8 (CH₂), 29.0 (CH₂), 51.5 (OCH₃), 63.3 (C), 172.5 (CO₂), 201.0 (CHO). - MS, m/z (%): 272 (1) [M⁺], 271 (4) [M⁺ - 1], 215 (44) [M⁺ - (CH₃)₂CH₃], 89 (100), 75 (46), 73 (55), 59 (30) [CH₃CO₂⁺], 57 (34). - C₁₄H₂₈O₃Si (272.5): calcd. C 61.72, H 10.36; found C 61.16, H 10.47.

(*R*)-(-)-Methyl 2-(*tert*-Butyldimethylsilyl)-2-formylheptanoate (**16d**): 0.33 g (0.80 mmol) of (*S,R*)-**15d** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.18 g (75%) of a colorless liquid; $[\alpha]_D^{20} = -126.1$ ($c = 0.93$, C_6H_6); ee = 98% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 3000$ -2800 cm^{-1} , 1710, 1690, 1465, 1430, 1410, 1390, 1365, 1230, 1190, 1110, 1040, 1005, 930, 875, 840, 820, 770, 675. - 1H NMR (CDCl₃): $\delta = 0.09$ [s, 3H, Si(CH₃)₂], 0.17 [s, 3H, Si(CH₃)₂], 0.86 (t, $J = 6.7$ Hz, 3H, CH₃), 0.92 [s, 9H, SiC(CH₃)₃], 1.00-1.35 (m, 8H, CH₂), 1.78-1.90 (m, 1H, CH₂), 2.01-2.11 (m, 1H, CH₂), 3.78 (s, 3H, OCH₃), 10.22 (d, $J = 1.7$ Hz, 1H, CHO). - ^{13}C NMR (CDCl₃): $\delta = -7.0$ [Si(CH₃)₂], -6.7 [Si(CH₃)₂], 14.0 (CH₃), 19.3 [SiC(CH₃)₃], 22.6 (CH₂), 27.1 [SiC(CH₃)₃], 26.7 (CH₂), 29.0 (CH₂), 29.8 (CH₂), 31.6 (CH₂), 51.5 (OCH₃), 63.3 (C), 172.5 (CO₂), 201.0 (CHO). - MS, m/z (%): 300 (1) [M⁺], 243 (100) [M⁺ - (CH₂)₃CH₃], 89 (61), 73 (31). - C₁₆H₃₂O₃Si (300.5): calcd. C 63.95, H 10.73; found C 63.85, H 10.82.

(*R*)-(-)-Methyl 2-(*tert*-Butyldimethylsilyl)-2-formyldecanoate (**16e**): 0.50 g (1.1 mmol) of (*S,R*)-**15e** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.27 g (75%) of a colorless liquid; $[\alpha]_D^{25} = -131.1$ ($c = 0.97$, C_6H_6); ee = 96% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 3000$ -2800 cm^{-1} , 1710, 1690, 1460, 1430, 1410, 1390, 1360, 1250, 1225, 1110, 1070, 1000, 930, 910, 835, 820, 770, 675. - 1H NMR (CDCl₃): $\delta = 0.09$ [s, 3H, Si(CH₃)₂], 0.17 [s, 3H, Si(CH₃)₂], 0.87 (t, $J = 7.1$ Hz, 3H, CH₃), 0.92 [s, 9H, SiC(CH₃)₃], 1.00-1.35 (m, 12H, CH₂), 1.78-1.89 (m, 1H, CH₂), 2.00-2.11 (m, 1H, CH₂), 3.78 (s, 3H, OCH₃), 10.22 (d, $J = 1.7$ Hz, 1H, CHO). - ^{13}C NMR (CDCl₃): $\delta = -7.0$ [Si(CH₃)₂], -6.7 [Si(CH₃)₂], 14.1 (CH₃), 19.2 [SiC(CH₃)₃], 22.7 (CH₂), 26.8 (CH₂), 27.1 [SiC(CH₃)₃], 29.0 (CH₂), 29.2 (CH₂), 29.3 (CH₂), 30.1 (CH₂), 31.8 (CH₂), 51.5 (OCH₃), 63.3 (C), 172.4 (CO₂), 201.0 (CHO). - MS, m/z (%): 329 (31) [M⁺ + 1], 313 (41), 183 (46), 89 (100), 81 (31), 73 (70), 59 (42), 57 (70), 55 (49), 44 (77), 43 (65). - C₁₈H₃₆O₃Si (328.6): calcd. C 65.80, H 11.04; found C 65.80, H 11.34.

(*S*)-(+)-Methyl 2-(*tert*-Butyldimethylsilyl)-2-formyldecanoate (**16e**): 0.23 g (0.50 mmol) of (*R,S*)-**15e** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.12 g (73%) of a colorless liquid; $[\alpha]_D^{28} = +107.6$ ($c = 1.22$, C_6H_6); ee = 98% (^{13}C -NMR analysis of the starting material). The spectroscopic data were in agreement with those of (*R*)-**16e**.

(*R*)-(-)-Methyl 2-(*tert*-Butyldimethylsilyl)-2-formyltridecanoate (**16f**): 0.37 g (0.80 mmol) of (*S,R*)-**15f** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.21 g (74%) of a colorless liquid; $[\alpha]_D^{21} = -112.1$ ($c = 1.00$, C_6H_6); ee = 96% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 3000$ -2800 cm^{-1} , 1710, 1690, 1460, 1430, 1410, 1390, 1360, 1250, 1230, 1115, 840, 820, 770, 675. - 1H NMR (CDCl₃): $\delta = 0.09$ [s, 3H, Si(CH₃)₂], 0.17 [s, 3H, Si(CH₃)₂], 0.88 (t, $J = 7.1$ Hz, 3H, CH₃), 0.92 [s, 9H, SiC(CH₃)₃], 1.00-1.35 (m, 18H, CH₂), 1.75-1.90 (m, 1H, CH₂), 2.00-2.11 (m, 1H, CH₂), 3.78 (s, 3H, OCH₃), 10.22 (d, $J = 1.7$ Hz, 1H, CHO). - ^{13}C NMR (CDCl₃): $\delta = -6.9$ [Si(CH₃)₂], -6.7 [Si(CH₃)₂],

14.1 (CH₃), 19.3 [SiC(CH₃)₃], 22.7 (CH₂), 26.8 (CH₂), 27.1 [SiC(CH₃)₃], 29.1 (CH₂), 29.4 (CH₂), 29.6 (CH₂), 29.7 (CH₂), 30.2 (CH₂), 32.0 (CH₂), 51.5 (OCH₃), 63.3 (C), 172.5 (CO₂), 201.0 (CHO). - MS, m/z (%): 370 (1) [M⁺], 314 (100), 313 (63), 253 (28), 89 (86), 75 (36), 73 (85), 59 (34). - C₂₁H₄₂O₃Si (370.65): calcd. C 68.05, H 11.42; found C 68.04, H 11.37.

(*S*)-(+)-2-(*tert*-Butyldimethylsilyl)-2-methylbutanoic Acid (**16g**): 0.72 g (2.3 mmol) of (*S,S*)-**15g** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.35 g (70%) of a colorless solid; m.p. 122-124 °C, $[\alpha]_D^{25} = +2.3$ ($c = 0.97$, C_6H_6); ee = 93% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 3500$ -3050 cm^{-1} , 2970, 2940, 2905, 2890, 2860, 1670, 1470, 1320, 1265, 1200, 1160, 1135, 860, 840, 825, 810, 775. - 1H NMR (CDCl₃): $\delta = 0.06$ [s, 3H, Si(CH₃)₂], 0.10 [s, 3H, Si(CH₃)₂], 0.88 (t, $J = 7.2$ Hz, 3H, CH₃), 0.95 [s, 9H, SiC(CH₃)₃], 1.23 (s, 3H, CH₃), 1.39 (d/q, $J = 13.8/7.2$ Hz, 1H, CH₂), 2.15 (d/q, $J = 13.7/7.3$ Hz, 1H, CH₂), 12.31 (br s, 1H, OH). - ^{13}C NMR (CDCl₃): $\delta = -7.2$ [Si(CH₃)₂], -7.1 [Si(CH₃)₂], 8.8 (CH₃), 16.4 (CH₃), 19.4 [SiC(CH₃)₃], 27.5 (CH₂), 27.7 [SiC(CH₃)₃], 37.8 (C), 185.1 (CO₂). - MS, m/z (%): 216 (1) [M⁺], 159 (25), 84 (24), 75 (100), 73 (36), 69 (20). - C₁₁H₂₄O₂Si (216.4): calcd. C 61.06, H 11.18; found C 61.08, H 11.14.

(*R*)-(+)-2-(*tert*-Butyldimethylsilyl)-2-ethylhexanoic Acid (**16h**): 1.04 g (2.94 mmol) of (*S,R*)-**15h** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.56 g (74%) of a colorless liquid; $[\alpha]_D^{25} = +9.0$ ($c = 0.92$, C_6H_6); ee = 85% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 3500$ -3050 cm^{-1} , 2960-2800, 2605, 1670, 1465, 1390, 1380, 1365, 1250, 1220, 830, 820, 810, 770, 735, 685. - 1H NMR (CDCl₃): $\delta = 0.08$ [s, 3H, Si(CH₃)₂], 0.09 [s, 3H, Si(CH₃)₂], 0.72-0.89 [m, 15H, CH₃, SiC(CH₃)₃], 1.12-1.24 (m, 4H, CH₂), 1.45-1.80 (m, 3H, CH₂), 1.87 (d/q, $J = 14.7/7.5$, 1H, CH₂). - ^{13}C NMR (CDCl₃): $\delta = -5.5$ [Si(CH₃)₂], -5.4 [Si(CH₃)₂], 10.1 (CH₃), 14.1 (CH₃), 19.7 [SiC(CH₃)₃], 23.7 (CH₂), 23.9 (CH₂), 27.6 (CH₂), 27.8 [SiC(CH₃)₃], 30.6 (CH₂), 42.1 (C), 184.5 (CO₂). - MS, m/z (%): 258 (2) [M⁺], 199 (26), 109 (23), 83 (50), 75 (100), 73 (59), 55 (26). - C₁₄H₃₀O₂Si (258.5): calcd. C 65.06, H 11.70; found C 64.67, H 11.47.

(*R*)-(+)-2-(*tert*-Butyldimethylsilyl)-2-ethyltetradecanoic Acid (**16i**): 0.70 g (1.5 mmol) of (*S,R*)-**15i** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.36 g (64%) of a colorless liquid; $[\alpha]_D^{21} = +11.6$ ($c = 1.09$, C_6H_6); ee = 88% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 3330$ -3050 cm^{-1} , 2960, 2930, 2860, 2620, 1675, 1465, 1390, 1380, 1360, 1260, 1250, 835, 820, 810, 770. - 1H NMR (CDCl₃): $\delta = 0.11$ [s, 6H, Si(CH₃)₂], 0.88 (t, $J = 7.1$ Hz, 3H, CH₃), 0.91 (t, $J = 7.1$ Hz, 3H, CH₃), 0.94 [s, 9H, SiC(CH₃)₃], 1.26 (m, 20H, CH₂), 1.64-1.90 (m, 3H, CH₂), 1.98 (d/q, $J = 14.6/7.4$, 1H, CH₂). - ^{13}C NMR (CDCl₃): $\delta = -5.50$ [Si(CH₃)₂], -5.46 [Si(CH₃)₂], 10.1 (CH₃), 14.1 (CH₃), 19.6 [SiC(CH₃)₃], 22.7 (CH₂), 23.8 (CH₂), 25.4 (CH₂), 27.7 [SiC(CH₃)₃], 29.4 (CH₂), 29.6 (CH₂), 29.69 (CH₂), 29.72 (CH₂), 30.7 (CH₂), 30.9 (CH₂), 32.0 (CH₂), 42.2 (C), 184.5 (CO₂). - MS, m/z (%): 370 (5) [M⁺], 299 (40), 157 (24), 83 (61), 75 (100), 73 (63), 55 (29). - C₂₂H₄₆O₂Si (370.7): calcd. C 71.28, H 12.51; found C 71.05, H 12.59.

(*S*)-(+)-2-(*tert*-Butyldimethylsilyl)-2-isopropyltetradecanoic Acid (**16j**): 0.96 g (2.0 mmol) of (*S,S*)-**15m** was oxidatively cleaved with ozone according to the procedure described for **4e** yielding 0.52 g (66%) of a colorless liquid; $[\alpha]_D^{25} = +8.7$ ($c = 0.98$, C_6H_6); ee = 93% (^{13}C -NMR analysis of the starting material). - IR (neat): $\tilde{\nu} = 3200$ -3060 cm^{-1} , 2960, 2930, 2860, 2620, 1675, 1470, 1390, 1370, 1265, 1255, 835, 820, 810, 770. - 1H NMR (CDCl₃): $\delta = 0.09$ [s, 3H, Si(CH₃)₂], 0.12 [s, 3H, Si(CH₃)₂], 0.88 (t, $J = 6.8$ Hz, 3H, CH₃), 0.96 [s, 9H, SiC(CH₃)₃], 1.03 (d, $J = 7.1$ Hz, 6H, CH₃), 1.21-1.78 (m, 22H, CH₂), 2.54 (sep, $J = 7.1$, 1H, CH). - ^{13}C NMR (CDCl₃): $\delta = -5.2$ [Si(CH₃)₂], -4.1 [Si(CH₃)₂], 14.1 (CH₃), 19.7 (CH₃), 20.0 [SiC(CH₃)₃], 22.1 (CH), 22.7 (CH₂), 27.5 (CH₂), 28.3 [SiC(CH₃)₃], 29.4 (CH₂), 29.6 (CH₂), 29.70 (CH₂), 29.72 (CH₂), 29.74 (CH₂), 31.5 (CH₂), 32.0 (CH₂), 47.0 (C), 183.8 (CO₂). - MS, m/z (%): 384 (8) [M⁺], 327 (38), 229 (23), 213 (47), 140 (28), 109 (22), 98 (43), 97 (81), 95 (29). - C₂₃H₄₈O₂Si (384.7): calcd. C 71.81, H 12.58; found C 71.58, H 12.60.

X-Ray Structure Determination of 12g¹⁹: Crystals of sufficient quality were obtained from diethyl ether/petroleum ether (1:1) at room temp. The compound crystallizes in the monoclinic space group P2₁, $a = 9.126(3)$, $b = 5.885(2)$, $c = 18.511(4)$ Å, $Z = 2$, $V = 991.35$ Å³, $M_r = 390.5$, resulting in a calculated density of $\rho_{calcd.} = 1.308$ $g\ cm^{-3}$, while the total number of electrons per cell amounts to $F(000) = 412$. $\sin \theta / \lambda_{max} = 0.65$ for solution and refinement. The structure was solved by direct methods as implemented in the Xtal3.0 package of crystallographic programs²³, employing GEN-SIN²⁴ to generate structure-invariant relationships and GENTAN²⁵ for the general tangent-phasing procedure. A total number of 5653 reflections was collected in the range $\pm h + k + l$, $\pm h - k - l$ at 0 °C with an Enraf-Nonius CAD4 diffractometer, $R_{av} = 0.012$, graphite-monochromated Mo-K α ($\lambda = 0.71069$ Å), $\mu = 1.86$ cm^{-1} , no absorption correction. 3435 reflections with $I > 2\sigma(I)$ were used in the full-matrix least-squares refinement process of

254 variables terminating at $R = 0.053$ [$R_w = 0.046$, $w = 1/\sigma^2(F)$] with a final shift/error smaller than 0.0013. Residual electron density 0.3. 15 hydrogen positions were located in a difference Fourier map and 7 hydrogen atomic coordinates were calculated by employing idealized geometric parameters. One methoxy group of the molecule is disordered in that carbon atom C19 occurs in two positions with almost equal site occupation parameters (C19A: 0.567, C19B: 0.43).

X-Ray Structure Determination of 16g^[9]: Crystals of sufficient quality were obtained from diethyl ether at room temperature. The compound crystallizes in the orthorhombic space group $P2_12_12_1$, $a = 10.849(2)$, $b = 11.717(1)$, $c = 10.943(1)$ Å. $Z = 4$, $V = 1391.04$ Å³, $M_r = 216.4$, resulting in a calculated density of $\rho_{\text{calcd}} = 1.033$ g cm⁻³, while the total number of electrons per cell amounts to $F(000) = 480$. $\sin \theta / \lambda_{\text{max}} = 0.631$ for solution and refinement. The structure was solved by direct methods as implemented in the Xtal2.6 package of crystallographic programs^[26], employing GENSIN^[27] to generate structure-invariant relationships and GENTAN^[28] for the general tangent-phasing procedure. A total number of 5787 reflections was collected in the range $+h +k +l$, $-h -k -l$ at -10°C with an Enraf-Nonius CAD4 diffractometer, $R_w = 0.0147$, graphite-monochromated Mo- K_α ($\lambda = 0.71069$ Å), $\mu = 1.43$ cm⁻¹, no absorption correction. 3862 reflections with $I > 2\sigma(I)$ were used in the full-matrix least-squares refinement process of 127 variables terminating at $R = 0.070$ [$R_w = 0.059$, $w = 1/\sigma^2(F)$] with a final shift/error smaller than 0.00026. Residual electron density 0.3. The positions of the hydrogen atoms were calculated and held fixed during the refinement process.

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[96/94]